

Product Data Sheet

Uniform Particle Size, Gel, Strong Base Anion Exchange Resin for Industrial Demineralization Applications

Description	DuPont [™] AmberLite [™] HPR4200 OH Ion Exchange Resin is a high-quality resin for use in industrial demineralization applications when high performance, high purity water, and cost-effective operation is required. The chemical properties and particle size of the resin have been balanced to combine excellent operating capacity with low pressure drop, while reducing chemical regenerant and rinse water usage.
	AmberLite [™] HPR4200 OH is compatible with all system technologies; it has the flexibility to be used in lead single or layered anion bed and in mixed bed polishers, allowing users to inventory only one strong base anion resin for their demineralization needs. In mixed bed applications, the light color of this anion resin is designed to allow easy visual distinction from the dark-colored cation resin following backwash separation.
	AmberLite [™] HPR4200 OH offers a quick start-up in a single bed or a mixed bed or when paired with weak base anion AmberLite [™] HPR9500 Ion Exchange Resin in layered bed systems. It can also be paired with weak base anion AmberLite [™] HPR9600 Ion Exchange Resin.
Resin Pairings	 Recommended pairing in industrial demineralization applications: AmberLite™ HPR1200 H Ion Exchange Resin (gel) – for mixed bed AmberLite™ HPR1300 H Ion Exchange Resin (gel) – for mixed bed AmberLite™ HPR9500 Ion Exchange Resin (macroporous) – for layered bed AmberLite™ HPR9600 Ion Exchange Resin (macroporous) – for layered bed
Applications	 Demineralization Ideally when treating water with: High percentage of silica When the treatment goal is:
System Designs	Compatible with all system technologies and bed configurations: • Co-current • Counter-current / Hold-down • Layered beds • Packed beds • Mixed beds

Historical Reference

DuPont[™] AmberLite[™] HPR4200 OH Ion Exchange Resin has previously been sold as DOWEX MARATHON[™] 4200 OH Ion Exchange Resin.

Typical Properties	Physical Properties		
	Copolymer	Styrene-divinylbenzene	
	Matrix	Gel	
	Туре	Strong base anion, Type I	
	Functional Group	Trimethylammonium	
	Physical Form	Yellow, translucent, spherical beads	
	Chemical Properties		
	Ionic Form as Shipped	OH-	
	Total Exchange Capacity	≥ 1.00 eq/L (OH ⁻ form)	
	Water Retention Capacity	60.0 – 66.0% (OH ⁻ form)	
	Ionic Conversion		
	OH-	≥95%	
	Particle Size [§]		
	Particle Diameter	730 ± 50 μm	
	Uniformity Coefficient	≤ 1.25	
	< 300 µm	≤0.3%	
	> 850 μm	≤ 10.0%	
	Stability		
	Whole Uncracked Beads	≥90%	
	Swelling	$CI^{-} \rightarrow OH^{-}$: 20%	
	Density		
	Particle Density	1.06 g/mL	
	Shipping Weight	655 g/L	
	[§] For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 45-D00954-en).		
Suggested	Temperature Range		
Operating	OH- form [‡]	5–60°C (41–140°F)	
Conditions	CI- form	5–100°C (41–212°F)	
	pH Range		
	Service Cycle	1 – 14	
	Stable	0-14	

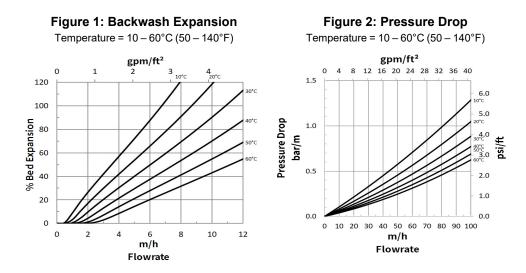
⁺ Operating at elevated temperatures, for example above 60 – 70°C (140 – 158°F), may impact resin life. Contact our technical representative for details.

For additional information regarding recommended minimum bed depth, operating conditions, and regeneration conditions for <u>mixed beds</u> (Form No. 45-D01127-en) or <u>separate beds</u> (Form No. 45-D01131-en) in water treatment, please refer to our Tech Facts.

Hydraulic Characteristics

Estimated bed expansion of DuPont[™] AmberLite[™] HPR4200 OH Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.

Estimated pressure drop for AmberLite[™] HPR4200 OH as a function of service flowrate and temperature is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean water.



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Please be aware of the following:

• WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

Have a question? Contact us at:

www.dupont.com/water/contact-us

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